

160 PU Ph D Nano Sciences and Technology

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Lorentz and Fitzgerald put forth the suggestion that there was contraction of bodies:-

- Along the direction of their motion through the earth
- Along the direction of their motion through the sun
- Perpendicular to the direction of their motion through the earth
- None of these

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Sound travels 40 m during 20 vibrations its wavelength λ is:-

- 0.5 m
- 2 m
- 4 m
- 3 m

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"A Moving particle, whatever its nature has wave properties associated with it." is known as:-

- De-Broglie hypothesis
- Bragg's hypothesis
- Frank's hypothesis
- None of these

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Water rises through a height h in a capillary tube of internal radius r . If T is the S.T. of water, then the pressure difference between the liquid level in the container and the lowest point of the concave meniscus is:-

- T/r
- r/T
- $r/2T$
- $2T/r$

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A gas expands adiabatically at constant pressure such that its temperature T is $1/\sqrt{2}$. The value of C_p/C_v of the gas is:-

- 2.00
- 1.50

- 1.30
- 1.67

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The ratio n of the velocity of the aircraft to the velocity of sound is referred to as:-

- Mach Number
- Reynolds Number
- Critical Number
- None of these

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The flow of heat from a hot body to a cold body is an example of:-

- Irreversible process
- Adiabatic process
- Reversible process
- Isothermal process

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A pendulum suspended from the roof of a train has a period T When the train is at rest). When the train is accelerating with a uniform acceleration 'a', the time period of the pendulum will:-

- Increase
- Decrease
- Remain unaffected
- Become infinite

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A body at higher temperature T in Kelvin) radiates heat at a rate which is proportional to:-

- T^4
- T
- T^{-4}
- T^2

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The addition of any velocity to the velocity of light merely reproduces:-

- Greater than the velocity of light

- The velocity of sound
- The velocity of light
- None of these

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The temperature of a black body is gradually increased. The colour of the body will change from:-

- White-green-red
- Yellow-green-red
- Red-violet-yellow
- Red-yellow-blue

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Which of the following experiment is a direct evidence for the quantised nature of energy states in atom?

- Frank Hertz experiment
- Fermi Dirac experiment
- Stern-Gelarch experiment
- None of these

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The resistance of a conductor is 5Ω at 100°C . What is its resistance at 0°C ?

- 4Ω
- 3Ω
- 2Ω
- 1Ω

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Moseley's law relates:-

- Frequency and atomic number
- Wavelength and intensity of X-Rays
- Wavelength and angle of scattering
- Frequency and Voltage applied

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The property of rotating the plane of vibration of a plane polarised light is called:-

- Optical photometry

- Optical activity
- Optical Illumination
- None of these

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In a Joule-Thomson experiment (Throttling process) :-

- The inversion temperature is the same for all real gases
- The inversion temperature is independent of the density of the real gas
- Ideal gases cannot be cooled for any P and T values
- Ideal gases can be cooled for certain P and T values

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A particle of mass m is moving with a constant velocity along a line parallel to the positive direction of the X-axis. The magnitude of its angular momentum w.r.t the origin:-

- Remains constant for all positions of the particle
- Goes on decreasing as x is increased
- Goes on increasing as x is increased
- Is zero

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A spherical solid ball of a kg mass and radius 3 cm is rotating about an axis passing through its centre with an angular velocity of 50 radian/s. The kinetic energy of rotation is:-

- 4500 J
- 910 J
- 9/20 J
- 90 J

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The spin angular momentum of an electron is:-

- In integral multiples of $h/2\pi$
- Always the same, $h/2\pi$
- In half integral multiples like $(s+1/2)*h/2\pi$ where S as running integer
- Always the same, $h/4\pi$

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A resistance potentiometer is a _____.

- Second order instrument
- First order instrument
- Zero order instrument
- None of the above

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A polarizer used in dry cell is:-

- Sodium carbonate
- Manganese dioxide
- Lead sulphate
- Ammonium chloride

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Which metal is protected by a layer of its own oxide:-

- Ag
- Al
- Fe
- Au

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Among the following the molecule with highest dipole moment:-

- CHCl_3
- CH_3Cl
- CCl_4
- CH_2Cl_2

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In the Neptunium series: ${}_{94}\text{Pu}^{241} \rightarrow \text{Am} \rightarrow \text{Np} \rightarrow \text{Pa} \rightarrow {}_{94}\text{U}^{233}$:-

- $\alpha, \alpha, \beta, \beta$
- $\alpha, \beta, \alpha, \beta$
- $\beta, \beta, \alpha, \alpha$
- $\beta, \alpha, \alpha, \beta$

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There is a plenty of room at the bottom. This was stated by:-

- Issac Newton
- Eric Drexler
- Richard Feynman
- Albert Einstein

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A plot of $\log[A]$ vs time (t) gives a straight line with negative slope. The order of the reactions:-

- 3
- 1
- 2
- Zero

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The poisonous gas evolved in Bhopal gas tragedy:-

- CO
- Methyl isocyanate
- Potassium cyanide
- None of the these

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A bucky ball is a molecules consisting of _____ carbon atom.

- 60
- 75
- 50
- 100

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Which is 3D silicates:-

- Talc
- Quartz
- Asbestos
- All of the above

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What is graphene:-

- Thin film made from fullerenes
- A one atom thick sheet of carbon
- A software tool to measure and graphically represent nanoparticle
- New material made from carbon nanotube

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Which of the following ions has zero crystal field stabilization energy in octahedral field:-

- Ca^{2+} low spin
- Fe^{3+} low spin
- Cr^{3+} high spin
- Fe^{3+} high spin

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Phenol undergoes the Freidel-Crafts reaction to form mainly the:-

- m-derivative
- P-derivative
- O-derivative
- All the above

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Which ratio decides the efficiency of nanosubstances:-

- Pressure/volume
- Volume/weight
- Weight/volume
- Surface area/volume

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Graphene is a:-

- Wide band gap semiconductor
- Not a semiconductor but behaves like graphite
- A narrow bandgap semiconductor
- Gapless band semiconductor

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Who coined the work nanotechnology:-

- Sumiolijima
- Richard Feynman
- Eric Drexler
- Albert Einstein

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A TCO is a semiconductor which has:-

- Low electrical resistivity and low optical transparency
- High electrical conductivity and low optical transparency
- High electrical resistivity and high optical transparency
- High electrical conductivity and high optical transparency

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Which pair is incorrect:-

- TiO-nonstoichiometric solid
- AgBr-Frenkel defect
- UO₂-anion deficient structure
- CaTiO₃-pervoskite

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Which of the following is microwave inactive:-

- CO
- NO
- HCl
- Cl₂

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The symmetry in quasi crystals is:-

- 4 fold
- 3 fold
- 5 fold
- 6 fold

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According to Stefan-Boltzmann law, heat loss proportion to:-

- T
- T⁴
- T⁶
- T²

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If three identical dice are rolled, then probability that the same number appears on each of them is:-

- 1/36
- 1/18
- 3/28
- 1/6

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If $f: \mathbb{R} \rightarrow \mathbb{R}$ be a function satisfying $f(2x + 3) + f(2x + 7) = 2 \forall x \in \mathbb{R}$ then fundamental period of $f(x)$ is:-

- 8
- 2
- 4
- 16

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166 PU_2016_160_E

If $a = \log_{24} 12$, $b = \log_{36} 24$, $c = \log_{48} 36$, then value of $(1 + abc)$ is:-

- 2ac
- 2ab
- 0
- 2bc

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Functions f and g are given by $f(x) = 3x^2 - 1$ and $g(x) = x^2 + 2$. Find an expression for:-

- $3x^4 + 12x^2 + 11$
- $4x^2 + 1$
- $3x^4 + 5x^2 - 2$
- $9x^4 + 1$

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If A and B are two square matrices of order n and $AB = B$, $BA = A$, then $A^2 + B^2 = 2I$ holds true for the condition:-

- |A| and |B| are non-zero
- |A| ≠ |B| ≠ 0
- |A| = |B| ≠ 0
- |A| = |B| = 0

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If all the roots of equations $(a-1)(1+x+x^2)^2 = (a+1)(x^4+x^2+1)$ are imaginary, then range of 'a' is:-

- $(-\infty, -2]$
- $(2, \infty)$
- $(-2, 2)$
- $(-2, \infty)$

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If the binomial coefficients of three consecutive terms in the expansion of $(1+x)^n$ are in the ratio 1 : 7 : 42, then value of 'n' is:-

- 50
- 55
- 65
- 32

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Total number of non-negative integral solutions of $18 < X_1 + X_2 + X_3 \leq 20$, is given by:-

- 441
- 1245
- 685
- 1150

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Let complex numbers z_1 and z_2 satisfy the conditions $|z + 6i| = 2$ and

$|z - 4i| = \left(\frac{z - \bar{z}}{2i} \right)$ then minimum value of $|z_1 - z_2|$ is:-

- 2
- 4

- 6
- 8

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Let x, y be non-zero real numbers and the expression $x^{12} + y^{12} - 48x^4y^4$ is not less than 'k', then value of 'k' is equal to:-

- 2^{12}
- -2^8
- -2^{12}
- 2^8

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Let 'M' be a 3x3 matrix, where $MM^T = I$ and $\det(M) = 1$, then:-

- $\det(MI)$ is always zero.
- $\det(M - I) \neq 0$.
- $\det(M + I)$ is always zero.
- $\det(M + 2I) = 0$.

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If $y = f\left(\frac{2x-1}{1+x^2}\right)$ and $f'(x) = \sin^2 x$, then $\left.\frac{dy}{dx}\right|_{x=0}$ is:-

- $\sin^2(1)$
- $1 - \cos 2$
- $-2 \sin^2(1)$
- $1 + \cos(1)$

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If $xy = y(dx + ydy)$, $y(1) = 1$ and $y(x) < 0$, then $y(-3)$ is equal to:-

- 1
- 3
- 2
- 3

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Minimum value of function $f(x) = \max\{x, x+1, 2-x\}$ is:-

- 3/2
- 1
- 1/2
- 0

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If circle $x^2 + y^2 - 2x - 6y + 8 = 0$ meets the y-axis at 'A' and 'B', then circumcentre of ΔABC , where 'C' is the centre of circle, is given by:-

- (0, 3)
- (1/2, 3)
- (1, 1/2)
- (1/2, 5/2)

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If $f(x)$ and $g(x)$ are differentiable functions for all $x \in [0, 1]$ such that $f(0) = g(1) = 2$, $g(0) = 0$ and $f(1) = 6$, then there exists some value of $x \in (0, 1)$ for which:-

- $f(\alpha) = 2g'(\alpha)$
- $f(\alpha) = 3g'(\alpha)$
- $f(\alpha) = 4g'(\alpha)$
- $f(\alpha) = g'(\alpha)$

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If the point $P(a^2, a)$ lies in region corresponding to the acute angle between lines $2y = x$ and $4y = x$, then 'a' belongs to:-

- 2, 6
- 4, 8
- 4, 6
- 2, 4

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For coplanar points $A(\vec{a}), B(\vec{b}), C(\vec{c}), D(\vec{d})$, if $(\vec{a}-\vec{d}) \cdot (\vec{b}-\vec{c}) = (\vec{b}-\vec{d}) \cdot (\vec{c}-\vec{a}) = 0$

then point D for ΔABC is:-

- Circumcentre

- Incentre
- Centroid
- Orthocentre

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If the pair of angular bisectors of the lines $y^2 - 3xy + 2x^2 - 4x + 6y - 16 = 0$ forms a triangle with the line $3x + 4y = 12$, then the orthocentre of triangle is given by:-

- (5, 8)
- (10, 12)
- (12, 10)
- (8, 5)

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If a matrix A is Hermitian, its Eigen values are always:-

- zero
- Real
- Complex
- infinite

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Semiconductors have the conductivity in the range of (ohm.m) :-

- 10^8
- 10^{-8}
- 10^4
- 10^{-2}

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Minimum number of slip systems that must be operative during plastic deformation:-

- 4
- 5
- 6
- 3

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Anisotropy is shown by _____ materials.

- single crystalline
- amorphous
- glass
- polycrystalline

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In a fiber reinforced polymer composites, for a given fiber volume content, Young's modulus depends on the orientation of the fiber with respect to the applied load Which orientation of the fibers will give the maximum value of Young's modulus?

- transverse
- longitudinal
- random
- both transverse and longitudinal

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The structures formed by rapid quenching from its molten state is known as:-

- pyrites
- metallic glasses
- chalcogenides
- perovskites

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Which one of the following material property is of significance in shock absorber:-

- hardness
- corrosion resistance
- fatigue
- yield strength

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Highly sensitive piezoresistive materials are made from:-

- single crystalline Si
- polycrystalline Si
- amorphous Si
- nanocrystalline Si

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At smallest sizes, colour of the gold nanoparticles become:-

- pink
- yellow
- red
- colourless

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Important property to be considered for shock resisting steel is:-

- low tensile strength
- high corrosion resistant
- low hardness
- high toughness

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Fuel cells are used to _____.

- generate energy
- harvest solar energy
- store energy
- induce photoelectric effect

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Burgers vector is related to:-

- dislocation
- acceleration
- deceleration
- unit cell

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Perovskites have the general formula of:-

- $A^{2+}B^{2+}X^{2-}_3$
- $A^{3+}B^{3+}X^{2-}_3$
- $A^{2+}B^{4+}X^{2-}_3$
- $A^{2+}B^{2+}X^{2-}_2$

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Seebeck effect is used in:-

- Thermoelectricity
- Piezoelectricity
- Piezoresistivity
- Electrostriction

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Grain boundary area is higher for_____ material.

- amorphous
- single crystalline
- poly crystalline
- nano crystalline

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For piezoelectricity generation:-

- silicon is preferred
- materials with charge asymmetry in the unit cell is preferred
- nanocrystalline materials are preferred
- centro symmetric structures are preferred

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Residual stress is not measured by:-

- transmission electron microscopy
- substrate curvature method
- nanoindentation
- X-ray diffraction

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The phenomenon of the growth of smaller particles in sol to bigger particle is known as:-

- annealing
- Ostwald ripening
- normalizing
- sintering

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The conductivity value:-

- increases with temperature for semiconductor
- does not depend on mobility of charge carriers
- does not change with dopant nature
- increases with temperature for metals

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Mean free path for electronic conduction is higher in:-

- nanotube
- nanowire
- nanoparticle
- quantum dots

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The saturation magnetization diminishes gradually and then abruptly drops to zero at the temperature known as:-

- Curie
- Meissner
- Neel
- Hall

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The biosynthesis of both RNA and proteins is dependent upon the nucleotide sequence of:-

- tRNA
- DNA
- mRNA
- rRNA

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A cell to cell channel is made up of:-

- 24 connexin
- 12 connexin
- 14 connexin
- 10 connexin

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Which of the following eukaryotic cell lacks nucleus:-

- Nerve cell
- WBC
- RBC
- Platelets

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When a molecule of pyruvic acid is subjected to anaerobic oxidation there is:-

- Gain of 2 molecules of ATP
- Loss of 3 molecules of ATP
- Loss of 6 molecules of ATP
- Gain of 4 molecules of ATP

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Which of the following is a non-membranous organelle?

- Plastid
- Endoplasmic Reticulum
- Ribosome
- Mitochondrion

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Cell growth occurs during:-

- Interphase and Postmitotic growth
- Interphase
- Mitotic phase
- Postmitotic growth

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In human beings, which part shows the minimum increase in weight from birth to adulthood?

- Brain
- Fat
- Skeleton
- Muscles

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196 PU_2016_160_E

Who among the following proposed the hypothesis: the bodies of animals and plants are composed of cells and products of cells?

- Robert Hooke
- Theodore Schwann
- Darwin
- Rudolf Virchow

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G-protein is:-

- Tetrameric
- Bimeric
- Unimeric
- Trimeric

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The cell wall of plants are made up of fibrils which predominantly contain:-

- Glucose
- Proteins
- Phospholipids
- Polysaccharides

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Each ribosome consists of two unequal subunits composed of:-

- RNA and proteins
- Only RNA
- DNA and proteins
- RNA and carbohydrates

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Thylakoids in a plastid are placed one above the other like a stack of coins to form a:-

- Granum
- Crista
- Stroma
- Matrix

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Different types of haemoglobin are produced in different stages of human development. It is an example of:-

- Multiplegene family
- Split genes
- Repeated genes
- Gene replacement

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The acetyl groups for cytoplasmic fatty acid synthesis appear in the cytoplasm, as a result of the activity of:-

- Citrate synthetase
- Isocitrate dehydrogenase
- Thiolase
- Citrate lyase

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Microsatellite sequences are repeat units with base pairs:-

- 11-60
- 5-30
- 1-6
- 20-120

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The enzyme which converts glucose to glucose 6-phosphate is:-

- Glucose 6-phosphate
- Hexokinase
- Glucose synthetase
- Phosphorylase

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Life without air is:-

- free from oxidative damage
- reduction
- impossible

- anaerobic

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When ATP is converted into ADP it releases:-

- Energy
- Hormones
- Oxygen
- Enzymes

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Zymase is:-

- Enzyme complex
- Pyruvate dehydrogenase
- Acetaldehyde dehydrogenase
- Pyruvate decarboxylase

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Respiration is an:-

- Anabolic process
- Endergonic process
- Exothermic process
- Endothermic process